

#### Year 11

Subject and Year Group	Autumn 1 Year 11	Autumn 2 Year 11	Autumn 2 Year 11	Spring 1 Year 11	Spring 2 Year 11	Summer 1 Year 11
Topic/Unit to be studied	C8: Chemical analysis	C2: Structure, bonding, and properties C5: Energy changes	C3: Quantitative chemistry	C4: Chemical change part 2 - Salts and Electrolysis	Revision for examination	Revision for examination
Core Knowledge (Substantive knowledge)	Purity of substances and how to test for purity. Chromatography as an analytical technique in chemistry. Testing for gases. Testing for unknown ions and spectroscopic techniques (T) <b>RP6</b> - Separation of coloured compounds with paper chromatography <b>RP7</b> - Chemical tests to identify unknown ions (T)	Learning to represent types of bonding with dot cross diagrams. Forming links between the structure and properties of different compounds. Uses of polymers and nanotechnology (T) To observe, describe and calculate energy change in reactions in terms of exothermic and endothermic reactions. Chemical and fuel cells (T) <b>RP4-</b> Temperature change in solutions	Applying knowledge of atoms to calculate Mr and to balance equations. Amounts of substance as masses, concentrations, and gas volumes (T) Yield and atom economy (T)	Synthesis and name salts from neutralisation reactions. Build on knowledge of reactivity of metals to describe extraction by electrolysis. <b>RP1-</b> Preparation of a soluble salt <b>RP2-</b> Titration (T) <b>RP3-</b> Electrolysis of solutions	Students use this time to consolidate their learning and practise exam technique.	Students use this time to consolidate their learning and practise exam technique.

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## Curriculum Map: Science



Core Skills (Disciplinary Knowledge)	- Use appropriate techniques, apparatus, and materials to carry out chromatography and gas tests. - Apply	- Using models to develop understanding, and to help explain properties - Apply mathematical concepts to calculate SA:V ratios (T) - Identify anomalies	- Apply mathematical concepts to use and rearrange equations to calculate A <sub>r</sub> , M <sub>r</sub> masses, moles, concentration, % mass - Understand, use and convert SI units. - Use of decimals, and	<ul> <li>Use appropriate techniques, apparatus, and materials during laboratory work, paying attention to health and safety.</li> <li>Select, plan and carry out methods to make salts from acids and</li> </ul>	
	mathematical concepts to calculate Rf values. - Use of decimals, and significant figures in calculations for chromatography. - Present and record observations from gas tests and positive and negative ion tests.	and apply mathematical concepts to calculate means. - Present data using appropriate methods, including tables and graphs including those with two lines of best fit - Make predictions using scientific knowledge and understanding. - Select, plan and carry out the most appropriate types of scientific enquiries to test predictions, including identifying independent, dependent and control variables.	significant figures - Apply mathematical concepts to use and rearrange equations to calculate atom economy, % yield, gas volumes (T)	metals, acids and carbonates and acids and bases - Select, plan and carry out the most appropriate types of scientific enquiries to test predictions, including identifying independent, dependent and control variables for electrolysis of solutions. - Make and record observations and measurements using a range of methods; and evaluate the reliability of methods and suggest possible improvements. - Reading new equipment accurately within a titration (T)	
		<ul> <li>Make and record observations and measurements using a range of methods; and evaluate the reliability of methods and suggest possible improvements.</li> <li>Interpret data to draw accurate conclusions.</li> <li>Apply mathematical concepts to calculate bon</li> </ul>			

### Resilience

Responsibility

Reflectiveness



## Curriculum Map: Science

# Success for Everyone



Assessment	End of Unit	End of Unit				
	assessment	assessment	assessment	assessment	assessment	assessment
	(MCQ/short	(MCQ/short	(MCQ/short	(MCQ/short	(MCQ/short	(MCQ/short
	answer/long	answer/long	answer/long	answer/long answer)	answer/long	answer/long
	answer) with	answer) with	answer) with	with interleaved	answer) with	answer) with
	interleaved content	interleaved content	interleaved content	content from	interleaved content	interleaved
	from previous units	from previous units	from previous units	previous units	from previous units	content from
						previous units